

## ВНУТРЕННИЕ БОЛЕЗНИ

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NEW PROMISING AGENTS  
FOR PHOTODYNAMIC THERAPY

*O.V. Shevchenko, N.G. Plekhova,  
O.V. Korshunova, I.G. Tananaev, V.I. Apanasevich*

*The problem of oncological diseases has acquired a social and medical character. In this regard, the rapid development of new drugs and methods, as well as modification of existing methods in order to increase their effectiveness, is currently required. The paper aims to develop and study two promising agents for photodynamic therapy. We discussed a new binary complex based on the widely used photosensitizer chlorin E6 with europium, which can solve one of the main disadvantages of drugs for photodynamic therapy: it can be used to treat deeper tumors and act more effectively by transferring luminescence energy between europium and chlorin E6. Our other agent was copper-containing chlorophyllin. Chlorophyllin is a natural component, which is always promising in the development of new drugs. The optical and fluorescent parameters of the agents were determined using physicochemical methods; their ability to generate reactive oxygen species in the presence of two fluorescent probes was studied: 2', 7'-dichlorodihydrofluorescein and dihydrofluorescein. Thus, we showed the effectiveness of chlorophyllin in comparison with chlorin E6 by 1.92 times at 50 µg / ml E6 and showed the effectiveness of the binary complex even at the minimum concentration of 10 µg / ml. Further research in cell cultures will develop the ways of the possible use of new substances as photosensitizers.*

**Keywords:** chlorin E6; chlorophyllin; binary complex; Europium; red spectrum; reactive oxygen species; photodynamic therapy; oncology

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## НОВЫЕ ПЕРСПЕКТИВНЫЕ АГЕНТЫ ДЛЯ ФОТОДИНАМИЧЕСКОЙ ТЕРАПИИ

*О.В. Шевченко, Н.Г. Плехова,  
О.В. Коришунова, И.Г. Тананаев, В.И. Апанасевич*

*Проблема онкологических заболеваний приобрела социальный и медицинский характер. В связи с этим в настоящее время требуется быстрая разработка новых препаратов и методов, а также модификация существующих методов с целью повышения их эффективности. Данная работа направлена на разработку и изучение двух перспективных агентов для фотодинамической терапии. Мы рассмотрели новый бинарный комплекс на основе широко используемого фотосенсибилизатора хлорина Е6 с европием, который может решить один из основных недостатков препаратов для фотодинамической терапии: его можно использовать для лечения более глубоких опухолей и действовать более эффективно за счет передачи люминесценции между европием и хлорином Е6. Другим нашим агентом был медьсодержащий хлорофиллин. Хлорофиллин – это природный компонент, который всегда является перспективным при разработке новых лекарств. Оптические и флуоресцентные параметры агентов были определены с использованием физико-химических методов; была изучена их способность генерировать активные формы кислорода в присутствии двух флуоресцентных зондов: 2', 7'-дихлордигидрофлуоресцеина и дигидрофлуоресцеина. Таким образом, мы показали эффективность (1) хлорофиллина по сравнению с хлорином Е6 в 1,92 раза при 50 мкг/мл Е6 и (2) бинарного комплекса даже при минимальной концентрации 10 мкг/мл. Дальнейшие исследования на клеточных культурах позволят разработать пути возможного использования новых веществ в качестве фотосенсибилизаторов.*

**Ключевые слова:** *хлорин Е6; хлорофиллин; бинарный комплекс; Европий; красный спектр; активные формы кислорода; фотодинамическая терапия; онкология*

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### Introduction

Scientists predict that by 2040, the number of newly diagnosed cancers will increase by 1.5 times compared to 2020 and will amount to 26 million people [12]. The method of photodynamic therapy [PDT] is currently being widely

developed due to the expansion of scientific knowledge, the capabilities of the instrumental base, as well as the rapidly increasing statistics of cancer.

Medicinal products of natural origin are beginning to occupy a leading position in many countries. Historically, the first photodynamic agents were natural ingredients [8]. The photosensitizer [PS] chlorophyllin is a derivative of the known chlorophyll compound. The main structure of chlorophyll is the porphyrin ring, which is analogous to the structure of heme in hemoglobin. The phytol tail of the porphyrin ring makes it fat-soluble and insoluble in water. Chlorophyllin is a semi-synthetic mixture of sodium-copper salts derived from chlorophyll [6]. During chlorophyllin synthesis, the magnesium atom in the center of the ring is replaced with copper, and the phytol tail is lost. Unlike natural chlorophyll, chlorophyllin is soluble in water, which is one of the essential requirements for a photosensitizer [7]. Currently, the chlorin E6 preparation synthesized in the early 90s in the Russian Federation from cyanobacteria is known [3]; it is widely used in many countries, as well as modified in different ways to enhance the effectiveness of photodynamic therapy [1].

In our work, we study the prospect of using the synthesized binary complex of chlorin E6 with europium not only for photodynamics but also for radiotherapy due to the introduction of a heavy element into the structure. Europium is capable of luminescence when exposed to coarse radiation (gamma or bremsstrahlung), as well as the transfer of this energy to the photosensitizer chlorin E6. We assume that in this way, it is possible to implement more effective therapy, for example, the treatment of deep tumors.

### **Materials and Methods**

Thus, the paper aims to study the possibility of a new generation of photosensitizers (chlorophyllin and a binary complex based on chlorin E6) to act as promising agents for photodynamic therapy.

At this research stage, the objectives are as follows:

- Synthesize a stable binary complex based on chlorin E6 with europium and select the optimal conditions for its further preparation.
- Investigate the optical and fluorescent parameters of the binary complex and copper-containing chlorophyllin using physicochemical research methods.
- Investigate the possibility of generating reactive oxygen species by two possible photodynamic agents in the presence of fluorescent probes.

To fulfill the set goal and objectives, several conditions were used.

The binary complex of chlorin E6 with europium was obtained at pH 6.5 by mixing a solution of chlorin E6 (OOO Veta-Grand, Russia, concentrate 5 mg

/ ml) and  $\text{EuCl}_3$  (Sigma-Aldrich, USA, 100 mg / ml) in a molar ratio of 11.7 mmol / 2, 93 mmol, respectively. The synthesis was carried out in the absence of light with stirring on a magnetic stirrer for 6 hours. The resulting solutions were dialyzed to remove low molecular weight compounds and excess europium against distilled water for 3 days and freeze-dried. Chlorophyllin from alfalfa was obtained from a company (Sigma-Aldrich, USA).

Absorption spectra were obtained in the range of 300–700 nm on spectrofluorimeters Synergy H1 (BioTek, USA) and RF-5301 PC (Shimadzu, Japan) in phosphate-buffered saline (PBS) at pH 7.4.

Irradiation with a red spectrum was carried out through a semiconductor laser (Light and Life, Tomsk, Russia) with an output power of 32 mW at a light wavelength of 645 nm, giving an irradiation dose of 6 J/cm<sup>2</sup>.

The generation of reactive oxygen species [ROS] was noted in the presence of fluorescent probes 2',7'-dichlorodihydrofluorescein (Sigma-Aldrich, USA) or dihydrofluorescein (Sigma-Aldrich, USA) on Synergy H1 (BioTek, USA) under optimal conditions  $\lambda_{\text{ex}} / \lambda_{\text{em}}$  485/530.

## Results

The obtained absorption spectra demonstrate the presence of characteristic absorption maxima, analogous to chlorin E6, which is part of the structures of chlorophyllin and the binary complex (Fig. 1). This fact is fully consistent with the data presented in the literature [2].

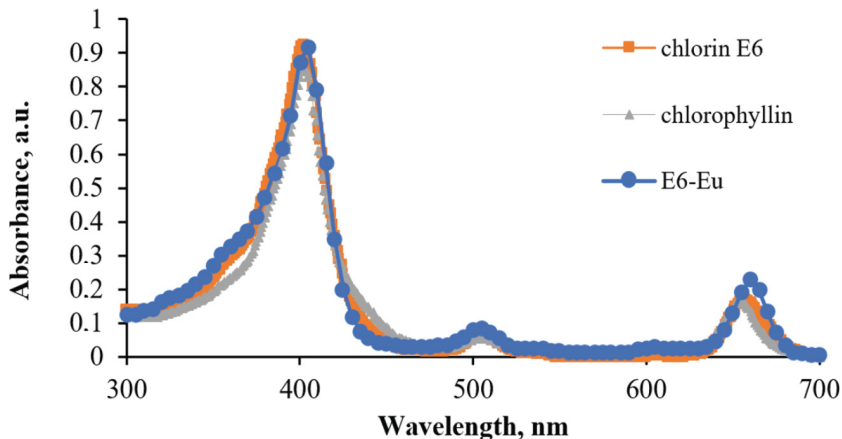


Fig. 1. Absorption spectra of E6-Eu, chlorin E6 and chlorophyllin at 50  $\mu\text{g/ml}$  E6;  $p < 0.05$ .

The E6 content was calculated at a characteristic absorption maximum of 400 nm since this wavelength is functionally significant during therapy [4]. This fact is important since modern LEDs with selective wavelength are used for photodynamic therapy.

In addition to optical fibers, dye lasers (600–700 nm), quantum scopes, semiconductor lasers (400–1270 nm), and various non-laser light sources are widely used [5].

Irradiation with the red spectrum of 645 nm was carried out for 30 min with an interval of 5 min in the presence of 4  $\mu\text{g/ml}$  fluorescent probe 2', 7'-dichlorodihydrofluorescein (Fig. 2). A significant increase in fluorescence has been shown for chlorophyllin compared to chlorin E6. In addition, the increase in fluorescence for a binary complex in 5 min is 1.5–2 times more intense.

Fig. 3 shows the generation of reactive oxygen species for chlorophyllin and chlorin E6 at an E6 content of 50  $\mu\text{g/ml}$ . At this concentration, the effectiveness of chlorophyllin at 30-min irradiation is 1.92 times.

Similar studies were conducted for the synthesized of binary conjugate under irradiation for 60 minutes. In this case, dihydrofluorescein was used as a fluorescent probe. Fig. 4 shows the experimental data at an E6 content of 10  $\mu\text{g/ml}$  for both photosensitizers.

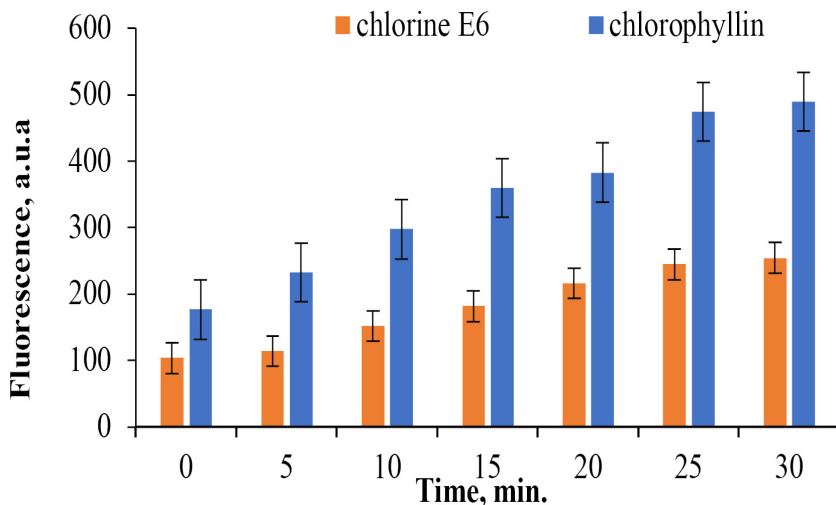


Fig. 2. An increase in fluorescence upon irradiation with red light in chlorin E6 and chlorophyllin at 80  $\mu\text{g/ml}$  E6 for 30 minutes;  $p < 0,05$ .

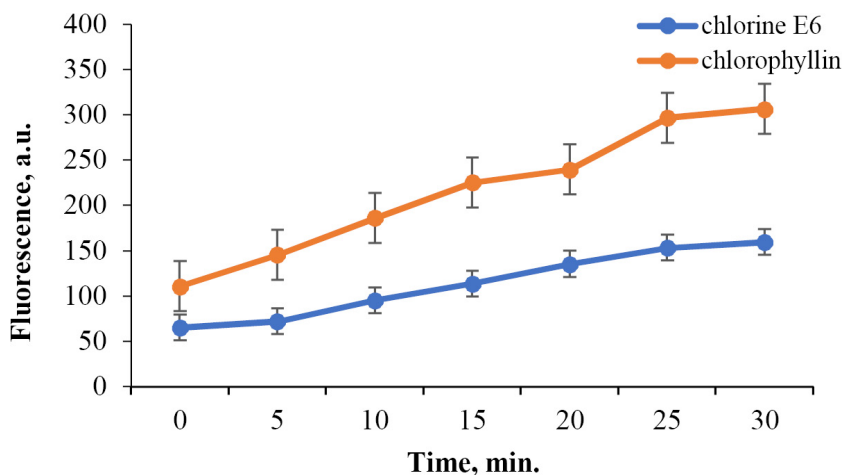


Fig. 3. Generation of reactive oxygen species by the test substances at 50  $\mu\text{g/ml}$  E6 in the presence of a fluorescent probe;  $p < 0,05$ .

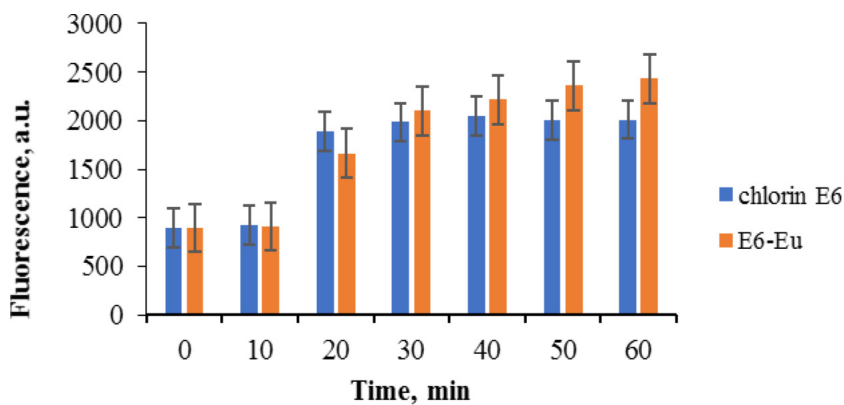


Fig. 4. Increase in fluorescence upon irradiation with a red light in chlorin E6 and E6-Eu for 60 min.

## Discussion

As mentioned earlier, one of the limitations of the photodynamic therapy method is the penetration depth of laser radiation into biological tissue. The drugs used in clinical practice are excited by light in the 620–690 nm region [11]. The permeability of biological tissues in this range is insignificant: only a few millimeters. The maximum tissue permeability is in the far red and near

infrared regions of the spectrum (750–1500 nm). That is, the absorption maxima of the main PS go beyond the tissue optical window (700–1100 nm), where most tissue chromophores, including hydroxy and deoxyhemoglobin, melanin, and fat have only weak absorption. The scholars [9] obtained hypericin derivatives. Hypericin has excellent photosensitizing properties and exhibits anti-tumor activity as well as minimal toxicity in the dark [10]. Potentially derived structures have an advantage over the parent compound due to photoactivation by red light, which penetrates deeper into the tumor tissue. The results show that three compounds (dibenzoxazole, pyridazinone and especially the dibenzthiazole derivative hypericin), capable of a bathochromic shift in the absorption spectrum, demonstrated efficient intracellular absorption and generation of singlet oxygen. In addition, they have a potent photocytotoxic effect under white light conditions.

However, the data also show that the present derivatives are unproductive photosensitizers when used under red light conditions, showing little generation of reactive oxygen species.

A binary complex synthesized by us as a PDT agent was also considered; it is worth noting its effectiveness in relation to the generation of ROS even at minimal concentrations. The copper-containing chlorophyllin and binary complex studied by us manifest themselves at the initial stages of the study as promising agents for PDT, the effectiveness of which exceeds the widely used chlorin E6.

### **Conclusion**

The heavy element europium introduced into the structures of the complex makes it possible to develop a combined method of radiophotodynamic therapy of deep malignant neoplasms. The data obtained confirm the relevance and perspective of further photodynamic studies of chlorophyllin and the binary complex.

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### **AUTHOR CONTRIBUTIONS**

The authors contributed equally to this article.

### **DATA ABOUT THE AUTHORS**

#### **Olga V. Shevchenko**

*Pacific State Medical University, Ministry of Health of Russia; Far Eastern Federal University*

*2, Ostryakov Ave., Vladivostok, 690002, Russian Federation; 8, Sukhanova Str., Vladivostok, 690091, Russian Federation*

*[tarakovaolga@gmail.com](mailto:tarakovaolga@gmail.com)*

*ORCID: <https://orcid.org/0000-0002-3113-3995>*

#### **Natalya G. Plekhova**

*Pacific State Medical University, Ministry of Health of Russia*

*2, Ostryakov Ave., Vladivostok, 690002, Russian Federation*

*[pl\\_nat@hotmail.com](mailto:pl_nat@hotmail.com)*

*ORCID: <https://orcid.org/0000-0002-8701-7213>*

#### **Oksana V. Korshunova**

*Pacific State Medical University, Ministry of Health of Russia*

*2, Ostryakov Ave., Vladivostok, 690002, Russian Federation*

*farmaoks@yandex.ru*

ORCID: <https://orcid.org/0000-0003-3533-8506>

**Ivan G. Tananaev**

*Far Eastern Federal University*

*8, Sukhanova Str., Vladivostok, 690091, Russian Federation*

*geokhi@mail.ru*

ORCID: <https://orcid.org/0000-0002-2159-8182>

**Vladimir I. Arpanasevich**

*Pacific State Medical University, Ministry of Health of Russia*

*2, Ostryakov Ave., Vladivostok, 690002, Russian Federation*

*oncolog222@gmail.com*

ORCID: <https://orcid.org/0000-0003-0808-5283>

**ДАННЫЕ ОБ АВТОРАХ**

**Шевченко Ольга В.**

*Тихоокеанский государственный медицинский университет  
Минздрава России; Дальневосточный федеральный университет  
пр. Острякова, 2, г. Владивосток, 690002, Российская Федерация;  
ул. Суханова, 8, г. Владивосток, 690091, Российская Федерация  
tarakovaolga@gmail.com*

**Плехова Наталья Г.**

*Тихоокеанский государственный медицинский университет  
Минздрава России  
пр. Острякова, 2, г. Владивосток, 690002, Российская Федерация  
pl\_nat@hotmail.com*

**Коршунова Оксана В.**

*Тихоокеанский государственный медицинский университет  
Минздрава России  
пр. Острякова, 2, г. Владивосток, 690002, Российская Федерация  
farmaoks@yandex.ru*

**Тананаев Иван Г.**

*Дальневосточный федеральный университет  
ул. Суханова, 8, г. Владивосток, 690091, Российская Федерация  
geokhi@mail.ru*

**Апанасевич Владимир И.**

*Тихоокеанский государственный медицинский университет  
Минздрава России  
пр. Острякова, 2, г. Владивосток, 690002, Российская Федерация  
oncolog222@gmail.com*

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